

# Physico-Chemical Properties of $\text{Ge}_{40}\text{Te}_{60-x}\text{Sb}_x$ Glassy Alloy

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## Abstract:-

We have systematically studied the physico-chemical properties of glassy  $\text{Ge}_{40}\text{Te}_{60-x}\text{Sb}_x$  system. Various chemical properties have been determined using different theories. Addition of Sb to  $\text{Ge}_{40}\text{Te}_{60}$  system was found to affect optical electronegativity, fractional ionic character and the degree of covalent character.

**Keywords:-** Optical electronegativity, Ionic and Covalent character.

## Introduction:-

The metal semiconductors occupy a unique place in material science not only due to their chemical, optical and thermal properties, but also due to their important role towards the advancement of technology [1-3]. Chemical properties of chalcogenide glasses have attracted significant attention because of their remarkable applications in radiating sensing of nuclear waste, mid-infrared Laser and many other technological fields [4-5]

Chalcogenide materials, especially those containing Ge and Te, are of importance in modern technology due to the following reasons: -

(1) Large change in optical constants between crystalline and amorphous state

(2) Rapid phase transition between amorphous and crystalline (two stable states) states. [6]

In view of the above, here we are concerned with the theoretical prediction of the compositional effect on the physico-chemical properties and how to correlate the different chemical parameters of  $\text{Ge}_{40}\text{Te}_{60-x}\text{Sb}_x$  ( $x = 0, 2, 4, 6$  and  $10$ ) glassy alloy.

## Theory and Discussion:-

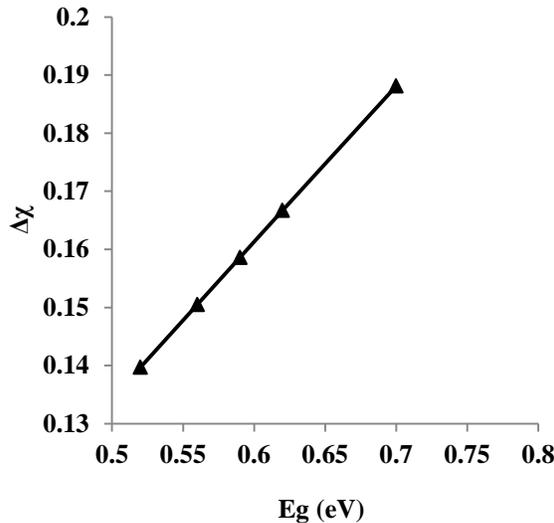
### [A] Optical Electronegativity:-

The optical electronegativity is an important parameter that relates to many physico-chemical parameters of the materials as suggested by Duffy [7]. The optical electronegativity  $\Delta\chi$  is the tendency of an atom or radical to attract electrons from the ionic bond. Duffy suggested a model that can be applied for all the compositions used in opto-electronic structure. The model correlates the relation between the energy band gap  $E_g$  and electronegativity  $\Delta\chi$  by the equation-

$$\Delta\chi = 0.2688 E_g \quad (1)$$

The obtained value of  $E_g$  has been reported in previous paper [8], and the value of  $\Delta\chi$  and  $E_g$

are listed in table-1. The variation of  $\Delta\chi$  with  $E_g$  are illustrated in Fig.-1. From the Fig.-1, it is shown that optical electronegativity decreases with decrease in energy band gap  $E_g$ .



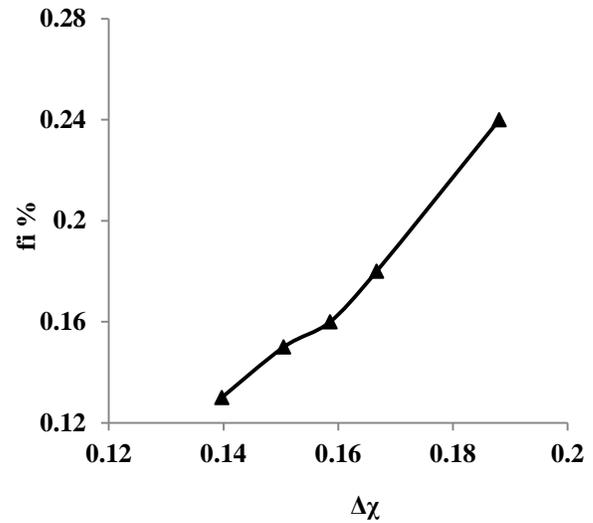
**Fig.-1:- Variation of  $\Delta\chi$  with  $E_g$**

**[B] Fractional Ionic Character:-**

Pauling [9] developed a thermo chemical method to formulate a reliable relationship between fractional character or ionicity of a bond and the difference in electronegativities of atoms forming the chemical bond. Following the theory due to Pauling, a relationship between fractional ionic character  $f_i$  and electronegativity difference  $\Delta\chi$  is considered as given below-

$$f_i = 1 - [\exp\{-(\Delta\chi)^2/4\}] \times 100\% \quad (2)$$

The calculated value of  $f_i$  is listed in table-1 and the variation of  $f_i$  with  $\Delta\chi$  is illustrated in Fig.-2 from the Fig.-2 it is clear that the variation of  $f_i$  decreases with the decrease in  $\Delta\chi$ .



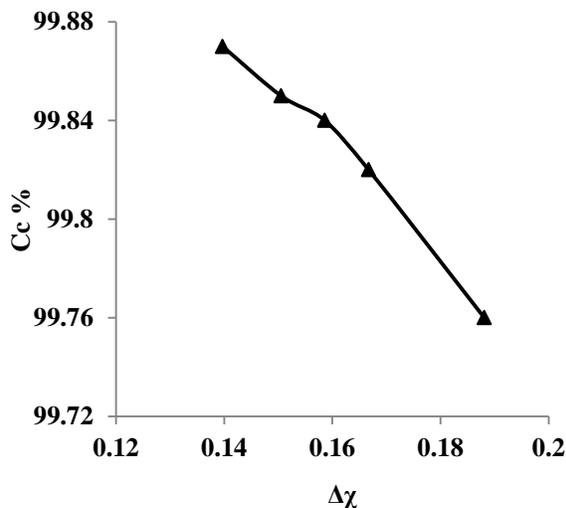
**Fig.-2:- Variation of  $f_i$  with  $\Delta\chi$**

**[C] The degree of covalence character:-**

Chalcogenide glasses are formed from materials like sulphur, selenium, arsenic, germanium, and tellurium which have predominantly covalent bonds; the degree of covalence in the bond of glassy system under investigation can be calculated by using the Pauling relationship [10] as given below-

$$C_c = 100 \exp[-(\Delta\chi/4)^2] \quad (3)$$

The calculated value of  $C_c$  is listed in table-1 and the variation of  $C_c$  with  $\Delta\chi$  is illustrated in Fig.-3. From the Fig.-3 it is shown that covalent character increases with the decrease in electronegativity.



**Fig-3: Variation of Cc with Δχ**

**Table-1: The value of  $E_g$ ,  $\Delta\chi$ ,  $f_i$  and  $C_c$  for the  $Ge_{40}Te_{60-x}Sb_x$  glassy system**

X at %	$E_g$ (eV)	$\Delta\chi$	$f_i$ %	$C_c$ %
$Ge_{40}Te_{60}$	0.70	0.1881	0.24	99.76
$Ge_{40}Te_{58}Sb_2$	0.62	0.1667	0.18	99.82
$Ge_{40}Te_{56}Sb_4$	0.59	0.1586	0.16	99.84
$Ge_{40}Te_{54}Sb_6$	0.56	0.1505	0.15	99.85
$Ge_{40}Te_{50}Sb_{10}$	0.52	0.1397	0.13	99.87

### Conclusion-

It is concluded here that the variation in Sb content in Ge-Sb-Te glassy alloys leads to change in the physic-chemical properties. As it is clear from various figures and table given that almost all the parameters, except the parameter  $E_g$ ,  $\Delta\chi$ , and  $f_i$  decreases with the increase in content of Sb in  $Ge_{40}Te_{60-x}Sb_x$  system. It has been found that  $C_c$  is proportional to composition x and increases with the increase in content of Sb from 0 to

10 at %, resulting in decrease of optical band gap and  $\Delta\chi$ .

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